### **IB Style Test - Topic 2: Molecular biology**

Includes new elements of the 2016 Guide



#### Multiple choice questions

1. Carbon is the basis of biological polymers because

I It can form chains by bonding to other carbon atoms
II It is found in carbohydrates
III It can form four covalent bonds

A I only

B I and II only

C I and III only

D I, II and III

2. When Wohler synthesised urea, it falsified which theory?

A Creationism

**B** Vitalism

C Spontaneous generation

D None of the above

3. Hydrogen bonding between water molecules is important in living systems because

I It makes water cohesive
II It gives water a high specific heat capacity
III It makes water transparent
IV It helps water to be a good solvent for polar molecules

A I and III only

B II and IV only

C I and II only

D I, II and IV only

4. Which row of the table shows a correct description of the different types of fatty acid?

|   | Saturated fatty acid     | Mono-unsaturated fatty acid | Poly unsaturated fatty acid |  |
|---|--------------------------|-----------------------------|-----------------------------|--|
| Α | Many double bonds        | One double bond between     | All the carbon atoms are    |  |
|   | between carbon atoms     | the carbon atoms            | linked by single bonds      |  |
| В | One double bond between  | Many double bonds           | Many single bonds           |  |
|   | the carbon atoms         | between carbon atoms        | between carbon atoms        |  |
| С | All the carbon atoms are | One double bond between     | Many double bonds           |  |
|   | linked by single bonds   | the carbon atoms            | between carbon atoms        |  |
| D | All the carbon atoms are | One single bond between     | Many double bonds           |  |
|   | joined by double bonds   | carbon atoms                | between carbon atoms        |  |



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- 5. Proteins are polymers with a large range of structures and functions because
  - A Each individual has a unique proteome
  - B They can be folded into different three dimensional shapes
  - C They are made of amino acids which are amphoteric
  - D They contain the element nitrogen
- 6. When a dipeptide is formed from two amino acids which type of reaction takes place?
  - A Denaturing of amino acids
  - **B** Condensation reaction
  - C Hydrolysis reaction
  - D A reaction which is catabolism
- 7. Which of the following are proteins?

I Collagen

II Glycogen

III Insulin

IV Immunoglobulins

A I and III only

B I, III and IV only

C I and II only

D All of the above

- 8. Crick and Watson elucidated the structure of DNA using which of the following?
  - A Model making
  - **B** Gel Electrophoresis
  - C X-ray crystallography
  - D Condensation reactions
- 9. Tag DNA polymerase could be used to
  - A Identify the DNA of a person at a crime scene
  - B Remove a portion of a DNA sample for testing
  - C Initiate transcription in a cell
  - D Produce multiple copies of a segment of DNA in laboratory conditions
- 10. The only N-containing nutrient available to a group of bacteria contains just heavy (15N) nitrogen molecules, All the bacteria divide just once and the new DNA in these new cells is examined (after DNA replication). What type of nitrogen would you find in that DNA?
  - A Only heavy nitrogen (15N) strands in the DNA molecules
  - B Some heavy (15N) and some light nitrogen strands in the DNA molecules
  - C One heavy (15N) and one light strand in each DNA molecule
  - D Variable nitrogen in the DNA molecules

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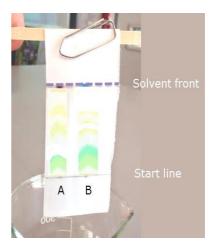


#### Structured answer questions

(4 marks)

11. Two chromatograms have been run to compare the pigments present in two angiosperm leaves, one which is green in colour and another which has a dark red appearance.

Describe how you could use the chromatogram to compare the pigments present in the two types of leaves including any measurements and calculations that you would make.



| 12. Why is the proteome of an individual cell of a multicellular organism different from proteome of the whole organism? | om the<br>(3 marks) |
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|--|-----------------------------------|
| 13. Describe how hydrogen bonding is important in water and in the structure                     | es of nucleic acids.<br>(4 marks) |
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| 14. Draw a generalised diagram of an amino acid molecule and label the cher make up the molecule | nical groups which                |
|  | (4 marks)                         |
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| 15. Lactose-free milk can be produced by the use of immobilised lactase. Explain why lacta hydrolysing enzyme and the importance of immobilising the enzyme in this process. |  |             |  |
|--|--|-------------|--|
|  |  | (5 marks)   |  |
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| 16   | Describe the role of yeast in the brewing and the baking industry. | (5 marks)   |  |
| 10.  | Describe the role of yeast in the brewing and the baking moustry.  | (5 IIIdIKS) |  |
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| 17. Discuss the reasons why lipids are important as long-term energy storage in animals. (3 marks) |
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| 18. Human insulin can now be produced in industry by bacteria.                                     |
| Explain why this fact demonstrates the universality of the genetic code. (2 marks)                 |
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